Biomorphous porous hydroxyapatite-ceramics from rattan (*Calamus Rotang*)

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Abstract The three-dimensional, highly oriented pore channel anatomy of native rattan (Calamus rotang) was used as a template to fabricate biomorphous hydroxyapatite $(Ca_5(PO_4)_3OH)$ ceramics designed for bone regeneration scaffolds. A low viscous hydroxyapatite-sol was prepared from triethyl phosphite and calcium nitrate tetrahydrate and repeatedly vacuum infiltrated into the native template. The template was subsequently pyrolysed at 800°C to form a biocarbon replica of the native tissue. Heat treatment at 1,300°C in air atmosphere caused oxidation of the carbon skeleton and sintering of the hydroxyapatite. SEM analysis confirmed detailed replication of rattan anatomy. Porosity of the samples measured by mercury porosimetry showed a multimodal pore size distribution in the range of 300 nm to 300 µm. Phase composition was determined by XRD and FT-IR revealing hydroxyapatite as the dominant phase with minimum fractions of CaO and Ca₃(PO₄)₂. The biomorphous scaffolds with a total porosity of 70-80% obtained a compressive strength of 3-5 MPa in axial direction and 1-2 MPa in radial direction of the pore channel orientation. Bending strength was determined in a coaxial double ring test resulting in a maximum bending strength of ~ 2 MPa.

1 Introduction

An increasing interest in the development of synthetic porous hydroxyapatite (HA) as a bone replacement material is caused by limited supply of autograft material and the health risks associated with the use of allografts [1]. Porous scaffolds for bone tissue engineering should exhibit mechanical properties similar to that of bone and meet certain criteria such as osseoinductivity and tailorable biodegradation [2, 3]. Porous hydroxyapatite ceramics were reported to show bioactivity and to induce bone-like apatite formation (e.g. carbonated hydroxyapatite) in vitro when exposed to simulated body fluid (SBF) [1, 4-7]. Thus, the implant may trigger accelerated healing and direct implant-to-bone bonding is facilitated [1]. High porosity and large pores enhance bone ingrowth and osseointegration of the scaffold once implanted in the body [8]. According to early work of Hubert et al. [9] the minimum pore size for a scaffold should exceed 100 µm. Subsequent studies have shown even more effective osseogenesis for pore sizes $>300 \,\mu\text{m}$ due to accelerated vascularization and oxygenation [8]. Over the past two decades many techniques have been developed to fabricate porous hydroxyapatite including polymeric sponge method [10], rapid prototyping techniques [11, 12], electrospinning [13], phase separation [14], particulate leaching [15], sacrificial filler [16], freeze casting [17], or gel-casting techniques [18]. Recently special attention has been paid to cellular scaffolds originating from biological tissue and materials like coral [19], cuttlefish [20], or bovine bone [21]. The aim of using biological structures as a template is to mimic natural structure hierarchy and anatomy at length scales covering a wide range from mm to nm [22]. The tissue of higher plants such as wood exhibits a hierarchical organisation of cells forming the vascular transportation system [3]. The honeycomb like microstructures of elongated hollow tubes may well be used for infiltration of liquid or gaseous reactants into the template. As described in literature, two different processes were developed to convert a template into a ceramic product: In the

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transformation process natural wood was pyrolysed to obtain biocarbon templates which were infiltrated with a metal melt or vapour and subsequently reacted to form carbide phases (SiC, TiC) [23, 24]. In the substitution process, native or pyrolysed wood templates were infiltrated with precursor solutions of inorganic salts or metal organic precursors. The carbon skeleton was removed by high temperature oxidation leaving an oxide replica of the template structure (Al₂O₃, ZrO₂, TiO₂, MnO) [25, 26]. In this work, highly porous hydroxyapatite scaffolds for bone tissue engineering were prepared from a natural plant tissue. Rattan plants of the subfamily Calamoideae of the family Arecaceae offer a pore channel pattern with a unique multimodal pore size distribution. Typical cell diameters range from ~ 50 nm to ~ 50 µm with long continuous pores (metaxylem vessels) up to 400 µm in diameter. The porosity structure of these scaffolds is similar to the anatomy of cortical bone with a porosity of 55–70% [27]. Thus, mimicking the plant anatomy offers an interesting potential for regeneration of cortical bone tissue.

2 Experimental

2.1 Hydroxyapatite sol synthesis

A sol was prepared from triethyl phosphite $(P(OC_2H_5)_3, 97\%)$, Fluka Chemie GmbH, Buchs, Swiss) and calcium nitrate tetrahydrate $(Ca(NO_3)_2 \cdot 4H_2O), \ge 99\%)$, Merck KGaA, Darmstadt, Germany) in ethanol/water solution. The molar ratio of water to the phosphorous precursor was kept at 8 for complete hydrolysis and the stoichiometric amount of the calcium to the phosphorous precursor at 1.67, respectively. The solution was allowed to age for 2 h at 60°C until a clear sol was obtained.

2.2 Template infiltration

Figure 1 shows the processing scheme applied. Cylindrical pieces of rattan (*Calamus rotang*) with a diameter of ~ 4 mm were dissected from internode sections of the stem. The pieces were cut into disks of 5–10 mm thickness and dried for 24 h at 105°C. The rattan samples were extracted with a mixture of toluol and ethanol at a volume ratio of 2:1 in a Soxhlet apparatus for 17 h to remove low-molecular weight species. Subsequently, samples were dried for 24 h at 105°C followed by a second extraction for 19 h using ethanol as solvent. After boiling the samples in distilled water for several hours a final drying for 24 h was proceeded at 105°C. For infiltration, the extracted samples were immersed in an evacuated glass beaker containing the hydroxyapatite-sol. After infiltration the samples were



Fig. 1 Flow chart for the preparation of porous hydroxyapatite ceramics via biotemplating from rattan plants

dried for several hours at 80°C allowing the sol in the samples to form a hydroxyapatite-gel. This procedure was repeated up to 3 times. The samples were pyrolysed at 800°C for 1 h in N₂-atmosphere. During pyrolysis the biopolymers (cellulose, polyoses and lignin) were decomposed leaving a porous carbon residue covered with poorly crystalline hydroxyapatite. Multiple infiltration processes were carried out to increase the hydroxyapatite content. Finally, the specimens were annealed up to 1,300°C in air to remove the carbon template by oxidation and to transform the sol–gel into crystalline hydroxyapatite.

2.3 Sample characterization

Microstructure and morphology of native rattan templates and template-derived HA-scaffolds were examined by SEM (Quanta 200, FEI, Eindhoven/Netherlands). For SEM-imaging the samples were mounted on sample carriers using conductive carbon cement (Wetzlar/Germany) and sputtered with gold. Observation by SEM was carried out in ultra high vacuum with an accelerating voltage of 10 kV. Pore size distribution was analysed by means of mercury intrusion porosimetry (Hg-Porosimeter 2000, Carlo Erba Instruments, Milan/Italy) assuming a spherical pore geometry. For reference, porosity of HA scaffolds was determined from the relative density which was calculated from the apparent density of the sample divided by the theoretical density of hydroxyapatite (3.156 g/cm³).

Chemical composition and crystalline phase content were analysed by XRD (Kristalloflex D500, Siemens, Karlsruhe/Germany) and FT-IR (Nicolet Impact 420, Nicolet Instruments, Madison/USA). Monochromated Cu-K α radiation ($\lambda = 0.014505$ nm, 30 kV, 30 mA) was applied for XRD of powdered samples. The crystalline



Fig. 2 Scheme of the double ring bending test arrangement

phases were identified according to the JCPDF files (HA: 09-0432, CaO: 37-1497, CHA: 19-272, β -TCP: 09-0169, α -TCP: 9-348 und TeCP: 25-1137). FT-IR-spectra were recorded in the range of 400–4,000 cm⁻¹ in transmission mode with a resolution of 1 cm⁻¹ using KBr technique. 1 mg of the sample powder was mixed with 300 mg KBr, grinded in a mortar and axially pressed into a transparent tablet.

The compressive and bending strengths of the porous hydroxyapatite scaffolds were measured (Instron 4204, Instron Corp., Canton/USA). Rectangular blocks with dimensions of $\sim 4 \times 4 \times 7$ mm³ were machined for compressive loading according to DIN V ENV 1291 at a crosshead speed of 1.0 mm/min. Samples were loaded in axial (*out-of-plane*) and radial (*in-plane*) directions of the pore channel orientation. A minimum of eight samples was tested to derive an average value of compressive strength. Bending strength was obtained from cylindrical disk specimen with diameters of ~ 30 –40 mm and a height of ~ 5 mm. Coaxial double ring test (DIN 52 292, Fig. 2) was performed with a radius ratio of load ring to support ring of r_1 : $r_2 = 4$ and a crosshead speed of 0.5 mm/min.

3 Results and discussion

3.1 Pore structure

Figure 3 shows the structure and morphology of the rattanderived hydroxyapatite ceramic sintered at 1,300°C as an exact replication of the native rattan tissue. Metaxylem vessels in the HA with diameters of $(289 \pm 15) \mu m$ are smaller than in the rattan template $(379 \pm 24) \mu m$ because of shrinkage during pyrolysis and sintering. Smaller phloem and protoxylem tracheids surrounding the large diameter metaxylem channels are well reproduced in the ceramic product and remained open. The grain size of HA sintered at 1,300°C ranges from 0.5 to 1.0 μm as determined from SEM micrographs. According to the results from mercury porosimetry, native rattan reveals lowest porosity of $\sim 55\%$, in comparison to HA ceramic with values of $\sim 60\%$ and carbon template of $\sim 68\%$. Porosities calculated from relative density (determined through geometrical dimensions) attain slightly higher values of 76% for native rattan, 78% for HA and 83% for biocarbon template. Pore size distributions of native rattan, pyrolysed rattan and biomorphous HA ceramics are shown in Fig. 4. A three-modal pore size distrubution can be detected for all samples: nanopores with pore sizes in the range of 10-100 nm refer to cell wall pores; pores with diameters of 10-100 µm correspond to the trachaidal cell lumae. The third class of pores in the range of 200-400 µm displays large xylem vessels in the template. After conversion into HA ceramic the class of smallest pores shows a broader distribution in size in the range of 300 nm to 8 µm than in native or pyrolysed rattan. Peaks in the range of 10-100 μ m and ~200 μ m can also be revealed in the HA ceramic which exhibits a similar three-modal pore size distribution as in the rattan template. Pore size distribution and porosity of HA ceramics lie within the range of spongy bone with pores from 200 to 400 µm in diameter and porosities of 55-70% [27, 28].

3.2 Phase composition

Figure 5 shows the FT-IR spectra of the rattan-derived hydroxyapatite scaffold sintered at 800-1,300°C. The main characterisitic bands of PO₄³⁻ together with characteristic modes of O-H groups confirm the presence of HA (Ca₁₀(PO₄)₆(OH)₂) in the reaction product. The peaks present at 567 and 604 cm⁻¹ correspond to the v_4 fundamental vibrational mode and arise from bending vibrations of the PO_4^{3-} ions. The peak at 960 cm⁻¹ is due to P–O symmetric stretching modes and is designated as the v_1 fundamental vibrational mode. Peaks at 1,036 and $1,086 \text{ cm}^{-1}$ arise from the v_3 fundamental vibrational mode and are due to the asymmetric stretching modes of the P–O bonds [29, 30]. The sharp absorption bands at 875, 1,420 and 1,460 cm^{-1} in the spectra sintered at temperatures of 800–900°C also indicate the presence of CO_3^{2-} ions but disappear in the spectra of samples sintered at higher temperatures [2]. XRD patterns (Fig. 6) show identical peaks to the characteristic patterns of stoichiometric HA (JCPDF 09-0432). Small fractions of CaO are indicated by the presence of relevant peaks (JCPDF 048-1467) in all of the samples. Additional peaks are due to small amounts of β -TCP (Ca₃(PO₄)₂, JCPDF 09-0169) at lower temperatures from 800°C to 1,100°C and α-TCP (JCPDF 09-0348) at higher temperatures of 1,200-1,300°C. However, no characteristic bands of β -TCP at 1,118 and 945 cm^{-1} were observed in the FT-IR spectra. According to literature, the presence of β -TCP at





temperatures above 700°C is due to thermal decomposition of non-stoichiometric or carbonated hydroxyapatite [31, 32]. At temperatures above 1,150°C, β -TCP transforms into α -TCP [33]. CaO is the major impurity in sol–gelderived hydroxyapatite [3] which is caused by insufficient aging of the precursor components mixture [34]. Calcium phosphates containing CaO are known to be biocompatible and bioresorbable [1, 35].

3.3 Strength and porosity

Figure 7 shows the compressive strength (*out-of-plane* loading) versus sintering temperature of the rattan-derived HA ceramics. A slight increase in fracture stress from

 (3.02 ± 0.53) MPa to (5.08 ± 1.66) MPa was observed with increasing sintering temperature from 800°C to 1,200°C. The biotemplated HA products exhibit porosity levels of 77–80%. Compared to human cancellous bone, which is characterized by a compressive strength of 5 MPa at porosities of 55–70%, the rattan derived biomorphous HA ceramics display similar strength levels when loaded along the pore channels [27]. Loading in radial direction, i.e. perpendicular to pore channels, however, revealed significantly lower values of (1.93 ± 0.85) MPa for samples sintered at 1,300°C.

Due to the similarity of pore channel morphology the mechanical behaviour of rattan derived ceramics may be described in analogy to honeycomb mechanics of brittle



Fig. 4 Distribution of pore size versus relative volume of pores obtained from mercury intrusion porosimetry for rattan, pyrolysed rattan and rattan-derived hydroxyapatite



Fig. 5 FT-IR spectra of biomorphous HA-ceramics sintered at temperatures from 800° C to $1,300^{\circ}$ C for 2 h

materials. In compression loading the cells suffer progressive crushing and in tension the pore channel structure fails by brittle fracture. For *in-plane* (_{ip}) compression loading e.g. loading stress acting perpendicular to the cell elongation, and for *out-of-plane* (_{op}) compression loading the fractional strength expressed by the ratio of crushing strength σ_{cr}^* to the failure strength of the strut material σ_{fs} is approximated by [36]

$$\frac{(\sigma_{cr}^*)_{ip}}{\sigma_{fs}} = \frac{4}{9} \left(\frac{t}{l}\right)^2 \quad \text{and} \quad \frac{(\sigma_{cr}^*)_{op}}{\sigma_{fs}} = 12 \left(\frac{t}{l}\right) \tag{1}$$

Equation 1 was derived for regular hexagonal cell morphology characterized by a strut thickness t and a strut length l (circumference 6 l). Since in the naturally



Fig. 6 XRD patterns of biomorphous HA ceramics sintered at temperatures from 800° C to $1,300^{\circ}$ C for 2 h



Fig. 7 Out-of-plane compressive strength versus sintering temperature of biomorphous HA-ceramics

grown tissue neither strut thickness *t* nor lumae dimension (*l*) remain constant but may locally vary significantly according to the seasonal growth conditions [37] the fracture stress as well as the other mechanical properties (Young's modulus, toughness) will vary in a periodical manner at least in the radial direction (seasonal growth direction). Even though Eq. 1 is a simplified approximation which does not take into account variation of *t* and *l* and the non-regular cell morphologies it can be used to estimate fracture properties of the strut material from the experimentally measured strength values. Approximating cylindrical cell morphology observed in rattan tissue by substituting $l \approx (\pi/3)r$ where *r* is the cylindrical pore channel radius the failure stress of the strut material is derived from

$$\sigma_{fs} \approx \frac{\pi}{36} \left(\frac{r}{t} \right) \left(\sigma_{cr}^* \right)_{op} \tag{2}$$



Fig. 8 SEM-micrograph of strut microstructure in porous HA-ceramic

for out-of-plane loading direction. Inserting experimental values for $(\sigma_{cr}^*)_{op}$ shown in Fig. 7 and assuming a reasonable strut thickness of $t = 5 \ \mu m$ a fracture stress of the struts σ_{fs} ranging from 2.1 to 21 MPa is derived for pore channel radii of 25 µm and 250 µm, respectively. These values are significantly lower compared to the data presented for hot-pressed HA of 560-700 MPa [29]. According to the common relations of strength and porosity of open cell foams [36] $\sigma_{cr}^*/\sigma_{fs} \approx 0.2(\rho^*/\rho_s)^{3/2}$ a porosity of 0.7-0.93 of the strut material would result in such low compression strength values. SEM micrograph analysis of a typical strut microstructure in the sintered HA, Fig. 8, however, indicates the porosity in the sintered struts to be less than 0.5. Regular development of intercellular spaces between stegmata and parenchyma cells may give rise for the formation of pores and inhomogeneities in the struts which are likely to cause low fracture stresses [38]. Re-infiltration of the strut is a possible way to increase strut density and hence raise σ_{fs} to achieve higher loading capacity of biomorphous HA ceramics derived from natural rattan template.

4 Conclusions

Biomorphous hydroxyapatite-ceramics with a porosity of 70–80% were fabricated by sol infiltration of a rattan template. After repeating sol infiltration in the carbon template, annealing in air at 1,300°C resulted in oxidizing of the carbon and sintering of the hydroxyapatite ceramic. The cellular anatomy of the native tissue template with a three-modal pore size distribution was transferred into HA ceramics. Despite of a high porosity, the HA scaffold

achieved sufficient mechanical strength to serve as spongious bone implants. Biomorphous HA ceramics fabricated in this study exhibit properties very similar to human cancellous bone in terms of porosity, pore size and mechanical stability.

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